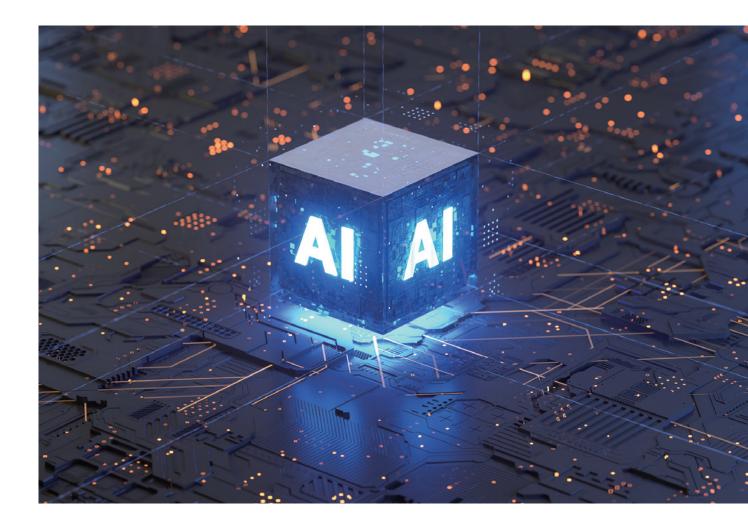


Solutions for Innovation

Scientific / Metrology Instruments Unknown Compounds Structure Analysis Software

msFineAnalysis Al

Automatic structure analysis software for data acquired by electron ionization and soft ionization Designed specifically for GC-HRTOFMS: JMS-T2000GC AccuTOF[™] GC-Alpha



JEOL Ltd.

Advanced Al technologies enable structure analysis of unknown compounds

msFineAnalysis AI offers a new structure analysis tool for unknowns that is specifically designed for the JEOL JMS-T2000GC "AccuTOF™ GC-Alpha." This next generation software adds this structure analysis capability to improve the overall automatic qualitative analysis functionality that was already available with our previous generation msFineAnalysis. The new **"integrated analysis"** combines GC/EI high resolution data, GC/soft ionization high resolution data, and **"structure analysis"** using two Als (Main AI, Support AI). These advanced AI technologies allow msFineAnalysis AI to provide a unique automatic structure analysis capability that was not previously available for GC-MS qualitative analysis.

.

P.9 Retention Index Qualitative Analysis



Differential Analysis

#4



Deconvolution Detection

Р.9

Group Analysis

Fragment lons
Additives
Off-flavor Components

Al Structure Analysis

Р.З-8

#

Integrated Analysis

· El/Soft Ionization Data Automatic Analysis

Library Database Search

· El Mass Spectrum Database Search

Analysis Using El Method Data Alone

· Molecular Ion Search in El Mass Spectrum

El Fragment Ion Analysis

· Accurate Mass Analysis



Molecular Formula Search
M/M+H Correction

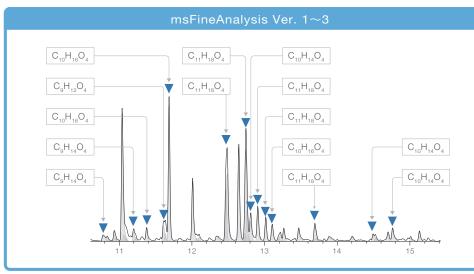


Evolving innovative solutions:

From molecular formula estimation to structural formula prediction for unknown compounds

For an unknown compound that is not registered in a library database (\mathbf{v}), the conventional msFineAnalysis algorithms automatically suggest a molecular formula. To take it a step further, msFineAnalysis AI enables automatic prediction of structures for all detected components.

Image of analysis results in the previous generation msFineAnalysis





Change by Al Structural Analysis

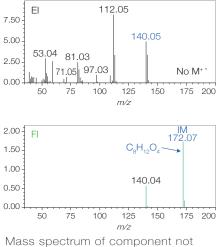
The necessity for soft ionization: Reliable acquisition of molecular formula information is the first step in structure analysis!

El mass spectral data is used for library databases so El methods are widely used for qualitative analysis of GC-MS samples. However, since El is a hard ionization method, many fragment ions are observed, and in many cases, it is not uncommon to observe minimal or no signal for the molecular ions.

Additionally, for unknown substances not registered in the library databases, it is difficult to distinguish, using the El mass spectra alone, whether the largest observed m/z is actually the molecular ion or just a fragment ion. In these cases, a soft ionization method is an effective tool for determining this information.

With the AccuTOFTM GC-Alpha, a variety of soft ionization methods including FI, PI, and CI are optionally available with the system. These techniques can assist in distinguishing ions (e.g. molecular ions and protonated molecules) that provide molecular weight information that then makes it possible to accurately determine the molecular formula information for unknown components.

Since molecular formula information is an important starting point for Al structure analysis, soft ionization is critically important for identifying unknown compounds.



registered in library

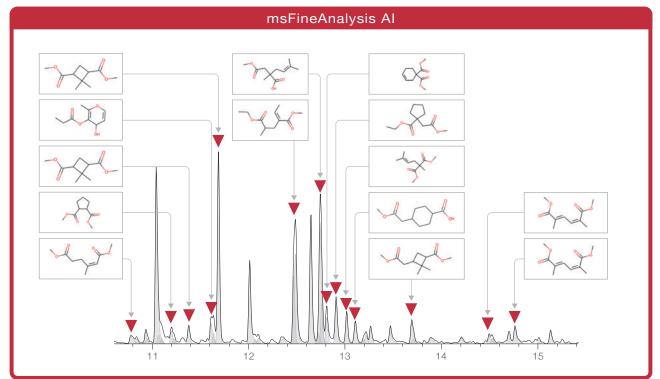
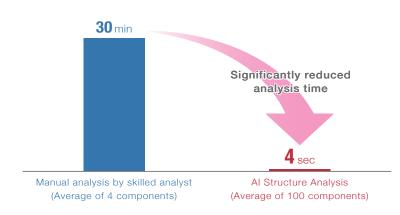


Image of analysis results in msFineAnalysis AI

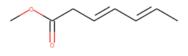
Manual Structure Analysis by Skilled Analyst vs. Al Automatic Structure Analysis



*Measured with JMS-T2000GC standard configuration PC

The time required for structure analysis was compared for the compounds observed in an acrylic resin measured by Py-GC-TOFMS and were not registered in the NIST library database.

Even an analyst with more than 30 years mass spectrometry experience required **approximately 2 hours for structure analysis of 4 components**, which is 30 minutes per component. On the other hand, AI structure analysis completed **100 components in less than 7 minutes**, which is 4 seconds per component.



Al Score: 779

Al structure analysis score (similarity) between the structural formula estimated by a skilled analyst and the correct structural formula, indicating that the structural formula is predicted with good similarity.

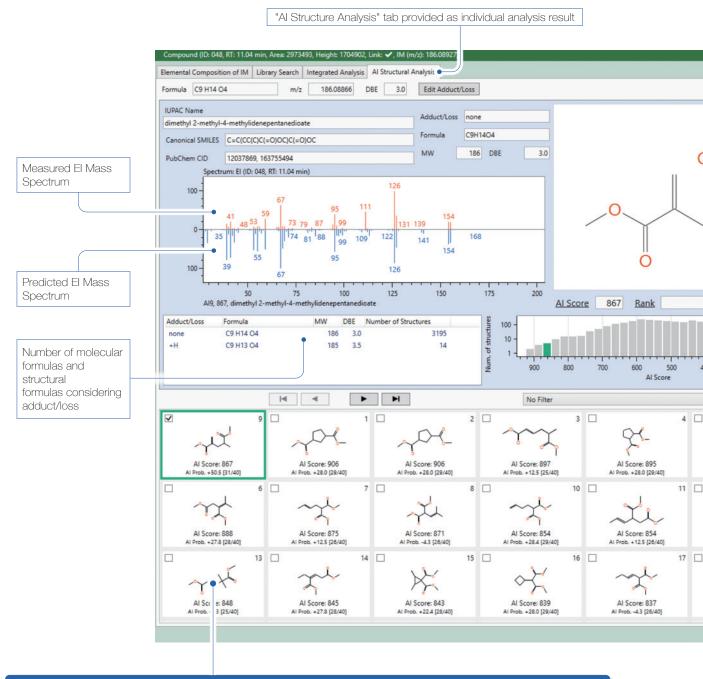


Automatic Structure Analysis Using Two Als:

Stable structure analysis without the need for an online environment

msFineAnalysis AI offers an automated structure analysis function.

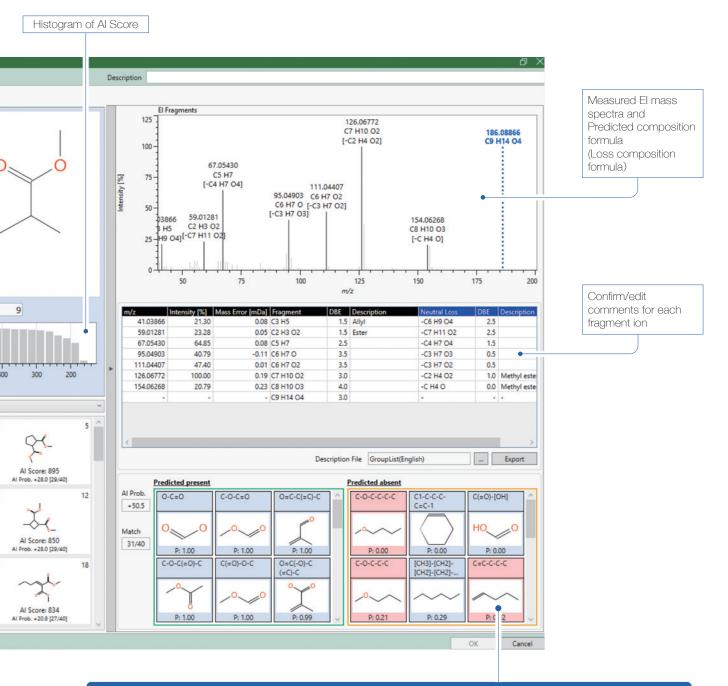
Based on the structural formula information of more than 100 million organic compounds recognized in the world and calculations using two newly developed AI models, it provides candidate structural formulas even for components that are not registered in the library database.



Structural Formula Prediction by Main Al

Determine a structural formula using mass spectra predicted from **100 million compounds** • Displays ranked structural formulas in list.

- \cdot Selecting a structural formula updates the information that is displayed.
- · Below each structural formula is an Al score that indicates the match percentage between structural formula and mass spectrum.



Partial Structure Prediction by Support AI

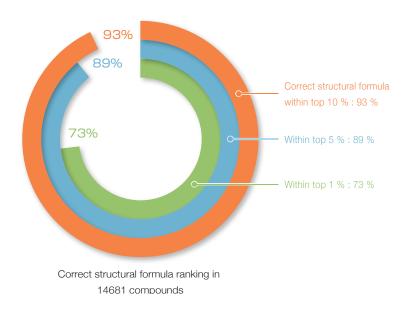
Assist in interpreting analysis result by **predicting partial structures** from the measured mass spectrum. • Displays the predicted partial structure information

- · Partial structure predicted and present are on the left, predicted and absent are on the right.
- Those with a blue background are the partial structures that match the selected structural formulas. Those with a red background are those that do not match.

AI Structure Analysis

Al Structure Analysis Prediction Accuracy

The NIST20 library database is used by the Main AI employed in AI structure analysis for learning and evaluation. The prediction accuracy tests confirmed that the correct structure was in the top 1% for 73% of the compounds, and in the top 10% for 93% of the compounds.



Structures with the same composition formula as the target compounds (14581 compounds) are prepared from PubChem. The rank order of the correct structure is verified among them.

Of the 14581 compounds verified, the correct structure was obtained within top 1% rank order in 73% (10644 compounds).

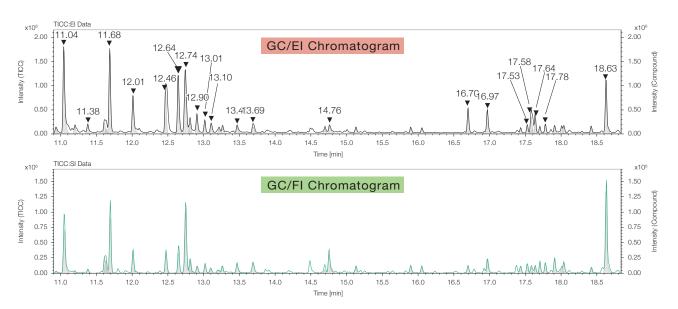
Subsequently, 6 known compounds that were not registered in the NIST20 library database were analyzed, and the correct structure was ranked highest for three compounds. For the remaining compounds, the top structures listed in the structure analysis results all shared similar features with the correct formula.

Compound Name	Correct Structure	Al Score	Rank Order (Total Candidates)		Тор	o 5 Structural Formu	las	
Cafenstrole	Loto	741	3 (2933)	AND	day.	Yart	76	×24-0
MCPA-thioethyl	$\rightarrow \phi$	735	1 (729)		¢.,	J.P.	¥-0-	9-0
Propaphos	40	802	1 (27)	ya	-ozu	42	1	yo
CNP-amino		710	14 (618)	ţa.	0¢	Ŷ¢.	06	J5-0-
Butamifos oxon	-}-	675	1 (56)	~ <u>}</u> ~\$	mo	- to	2 h	**
Isoxadifen-ethyl	70	586	22 (5348)	-fo	AG	78	AB	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~

: Correct structure

Al Structure Analysis Application

The AI structure analysis results for an acrylic resin measured by Py-GC-HRTOFMS are shown below. For the 20 compounds that had structures that could not be confirmed with the NIST20 library database, AI structure analysis was performed after determining their molecular formula through integrated analysis.



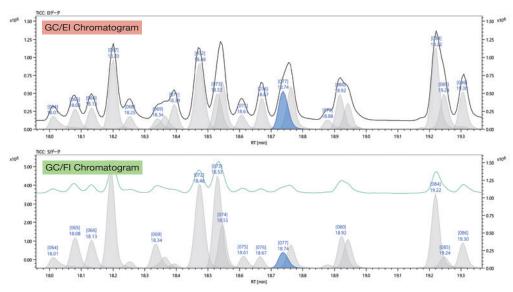
All of these compounds had structures that included methyl esters (reflecting the acrylic resin monomer structure) shown for the higher ranked possibilities. This information allowed us to estimate that all of the compounds targeted for structure analysis are acrylic resin thermal decomposition products.

RT	Composition Formula		Top 5 S	structural Fo	rmulas	
11.04	C9 H14 O4	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~	A	101	rte	re
11.38	C10 H16 O4	Fr	H	701	fit	$\sim \sim \sim$
11.68	C10 H16 O4	Fr	\sim	H	Hal	fat
12.01	C9 H14 O4	yor	rte	A	-	the
12.46	C10 H16 O4	H	Fr	ret	y h	fat
12.64	C10 H16 O4	of.	44	4t	ret	~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~~
12.74	C11 H18 O4	pth	-10	-up	the	Y
12.90	C11 H18 O4	3r	rol	24	Å	-Re
13.01	C11 H18 O4	Re	-3¢-	uge	201	the
13.10	C11 H18 O4	-04	Laf	yer	-the	No

RT	Composition Formula		Top 5 S	structural For	mulas	
13.46	C10 H14 O4	y	12	201	Fr.	p
13.69	C11 H18 O4	Yr	AL	24	to	hy
14.76	C10 H14 O4	ふど	-th		N	25
16.70	C13 H20 O6	zzr	mit	Lunite	र्भ्रद	ste
16.97	C13 H20 O6	zzr	mit	Jun'y	उद्गे	to the
17.53	C14 H22 O6	A	~Pric	sig	-ge	-Lec
17.58	C13 H20 O6	zzr	mit	र्भ्र	XYC	refine
17.64	C14 H22 O6	戒	- Lec	mile	A.	milie
17.78	C14 H22 O6	A	- Lec	zte	.ge	milie
18.63	C15 H24 O6	'too	-År	the	mqu	لمريد

#2 **Deconvolution Detection**

Chromatographic peak deconvolution can detect trace components that may not be obvious in the TICC due to the coelution of several components.

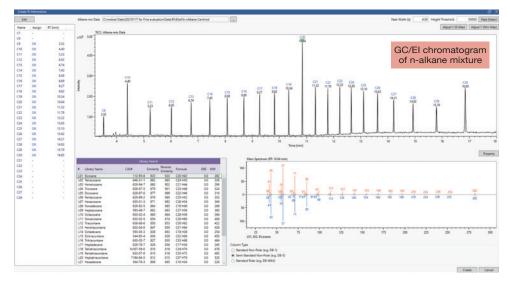


El: black solid line: TICC, gray peaks: deconvolution peak (blue: currently selected) Fl: green solid line: TICC, gray peaks: deconvolution peak (blue: currently selected)

This step simplifies the data analysis process by defining which ions go with each compound and eliminates the need for creating extracted ion chromatograms (EICs).

3 Retention Index Qualitative Analysis

Retention index (RI) is a relative index value based on the retention times (RT) for an n-alkane standard mixture.

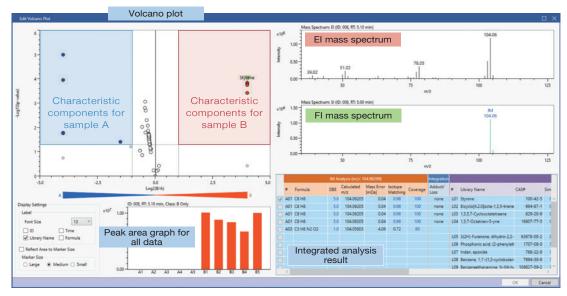


Retention Index Creation Screen

This qualitative analysis function is performed by converting the RT of the target component into a RI and then comparing it with the RI value listed in the databases, etc. With msFineAnalysis AI, it is possible to narrow down the qualitative analysis result further by using the RI.

Differential Analysis · Two Sample Comparison

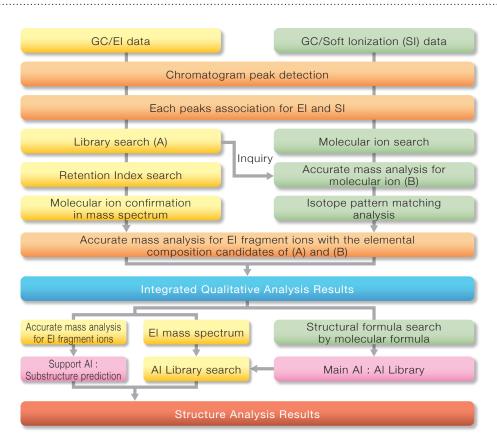
This function uses the reproducibility of the p-value on the vertical axis and a volcano plot which indicates the intensity ratio between two samples on the horizontal axis.



Detailed analysis – Volcano plot (A: Reference product, B: Defective product)

This information enables a visual confirmation of the differing components between two samples. For example, it is possible to confirm if a component increases or decreases when comparing a reference product to a defective product or to identify characteristic components in a new material by comparing it to an existing material. For a two sample comparison, it is possible to set n=1, 3, 5 for the number of measurements for each sample.

msFineAnalysis Al Analysis Flow



JMS-T2000GC Acc	uTOF™ GC-Alpha
Mass Resolution	30,000@ <i>m/z</i> 614
Mass Accuracy	1 ppm@El standard ion source
Mass Range	<i>m/z</i> 4-6,000
Ionization Methods	EI、CI、PI、FI、FD、DEI、DCI

msFineAnalysis Al	
Specifications	· Automatically detecting peaks and creating mass spectra
	· Creating mass spectra by manual peak detection
	· Creating mass spectra by deconvolution processing
	\cdot Analyzing identical components of two measurement data items
	\cdot Analyzing molecular ions according to two mass spectra
	· Variance component analysis
	· Displaying analysis results using retention indices
	· Displaying NIST database search results
	· Displaying exact mass calculation results
	· Displaying isotope pattern analysis results
	· Displaying measurement conditions
	· User interface: English
	· Al Structure analysis
	0

Certain products in this brochure are controlled under the "Foreign Exchange and Foreign Trade Law" of Japan in compliance with international security export control. JEOL Ltd. must provide the Japanese Government with "End-user's Statement of Assurance" and "End-use Certificate" in order to obtain the export license needed for export from Japan. If the product to be exported is in this category, the end user will be asked to fill in these certificate forms.

JEOL Ltd. 3-1-2 Musashing www.jeol.com IS

JEOL

3-1-2 Musashino Akishima Tokyo 196-8558 Japan Sales Division Tel. +81-3-6262-3560 Fax. +81-3-6262-3577 www.jeol.com ISO 9001 · ISO 14001 Certified



• AUSTRALIA & NEW ZEALAND • BELGIUM • BRAZIL • CANADA • CHINA • EGYPT • FRANCE • GERMANY • GREAT BRITAIN & IRELAND • INDIA • ITALY • KOREA • MALAYSIA • MEXICO • RUSSIA • SCANDINAVIA • SINGAPORE • TAIWAN • THAILAND • THE NETHERLANDS • UAE (Dubai) • USA