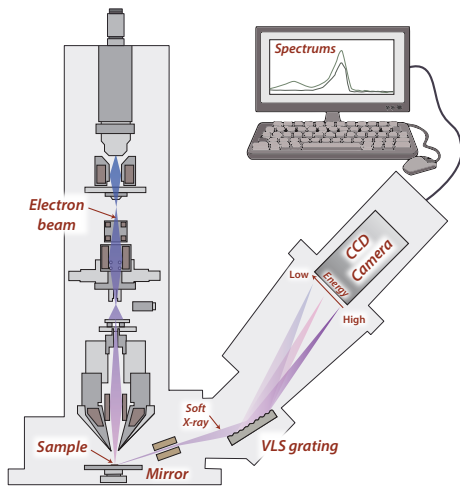


## Principle and features



Characteristic X-rays are generated by the electron beam from the sample and guided, by the focusing mirror, to the newly developed diffraction varied-line-spacing (VLS) grating. X-rays diffract from the grating according to their wavelength; different characteristic X-rays are detected simultaneously by a parallel detection system (CCD camera) and displayed as a spectrum.

### Benefits:

- ✓ **Extreme spectral resolution** → chemical state analysis of light elements
- ✓ **Excellent light element detection** (Li, Be, B, C, etc..)
- ✓ **High sensitivity** → Example: a few Boron ppm in steel can be detected
- ✓ **High stability and reproducibility**
  - No moving parts, possibility to switch between gratings

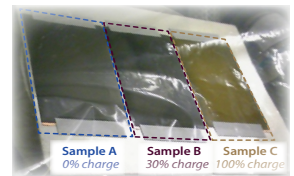
## Study case: Li K emission spectra, chemical bond effect

### Methodology

Ref: H Takahashi et al 2016 IOP Conf. Ser.: Mater. Sci. Eng. 109 012017

Three samples of a lithium-ion battery anode with different charging conditions are analyzed with SXES: 0 % (A), 30 % (B), and 100 % (C)

→ 3 maps constructed in 3 energy ranges corresponding to **satellite of Li-K**, **Li-K** and **C-K**

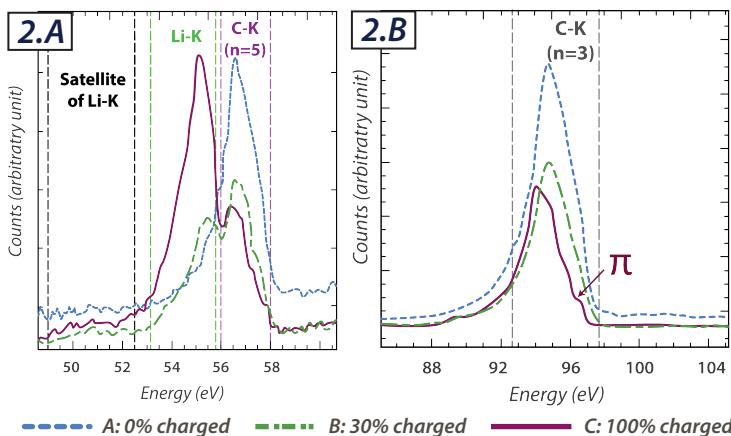
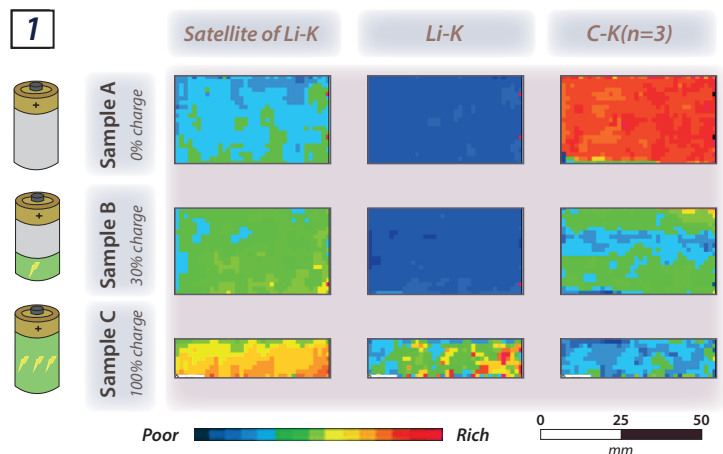


### Results: maps and spectra

**Li-K Satellite Map:** The map due to the *satellite of Li-K* (Fig 1) showed an increase in relative intensity with increasing charge of the anode material.

**Li-K Map:** The metallic lithium is formed locally and distributed heterogeneously under the fully charged condition.

**C-K(n=3) Map:** The map indicates the intensity distribution due to graphite; the other constituent of the anode. Under the fully discharged condition, the intensity was strongest. The intensity reduced substantially when the amount of charging increased by 30 % and stayed at this level when the amount of charging increased to 100 %.



**High resolution:** The spectra corresponding to the three charge states are extracted from the maps in Fig1. They cover all energy ranges of the *satellite Li-K*, *Li-K* and *C-K(n=5)* (Fig 2.A), as well as *C-K(n=3)* (Fig 2.B). Thanks to the **high energy resolution of the SXES**, the peaks of *Li-K* and *C-K(n=5)*, which have a very close energy, are clearly distinguished.

**Fast and accurate interpretation of results:** The discharged anode does not show any Li peak unlike the two charged anodes (Fig 2.A). On the contrary, the discharged anode shows the strongest C peaks, while the 100% charged anode shows the smallest peaks (Fig 2). Moreover, the intensity of the  $\pi$  sub-peak in the *C-K(n=3)* (Fig 2.B) represents the battery state of charge: the more pronounced the peak, the more charged the battery is and vice versa.

→ Thus, for each sample, the intensities of the Li and C peaks are representative of the battery's state of charge.

### In summary

High spectral energy resolution → ability to **distinguish the close energy peaks** between the *Li-K* emission and *C-K(n=5)*

**High potential** for the **characterisation of lithium** ion battery anodes

A wide area distribution of **chemical state spectral Li map** can be provided by EPMA/SEM-SXES